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| **Test Tracking** |
| **Objective Topic** | **Question #s** | **Points Earned** | **Points Possible** | **% Mastery** |
| 2.1: I will distinguish between pure elements and mixtures. | 16, 28 |  | 2 |  |
| 2.2: Particle Diagrams | 28, 29, 30 |  | 3 |  |
| 2.3: I will distinguish between compounds and mixtures. | 29, 30 |  | 2 |  |
| 2.4: I will distinguish between the two types of mixtures. | 15 |  | 1 |  |
| 2.5: Separating Mixtures | N/A | N/A | N/A | N/A |
| 2.6: I will distinguish among the different states of matter. | 21, 26a, 26b, 26c |  | 4 |  |
| 2.7: I will analyze the entropy for different states of matter. | 14, 17, 26d |  | 3 |  |
| 2.8: I will distinguish between heat and temperature.  | 27 |  | 1 |  |
| 2.9:  I will convert temperature between Celsius and Kelvin. | 25 |  | 1 |  |
| 2.10: Heating Curve | 1, 2, 20, 22 |  | 4 |  |
| 2.11: Cooling Curve | N/A | N/A | N/A | N/A |
| 2.12: Heat required for a phase change. | 4, 6 |  | 2 |  |
| 2.13: Heat required for a temperature change. | 3 |  | 1 |  |
| 2.14: I will apply heat equations to a heating curve. | 19 |  | 1 |  |
| 2.15: I describe the Kinetic Molecular Theory.  | 8, 9 |  | 2 |  |
| 2.16: I will analyze vapor pressure using Table H | 5, 7, 23 |  | 3 |  |
| 2.17: Pressure vs. Volume | N/A | N/A | N/A | N/A |
| 2.18: Temperature vs. Volume; Pressure vs. Temperature | 11, 12, 13, 24 |  | 4 |  |
| 2.19: Combined Gas Laws | 10, 18 |  | 2 |  |

**When you are done tracking, complete the reflection guide on back page 🡪**

**Did you get the score you expected? Is it higher or lower than you expected?**

Now that you’ve seen the breakdown of your score, it’s time to consider what you’ll do to increase your mastery of these topics. As a student, how can you ensure that you are doing and showing your best each day? Answer the following reflective questions, based on the information above, and create the **goals** that will make up your **Action Plan**.

**If YES!, describe what actions contributed to your success. How did you meet your goal?**

**If NO!, describe the actions that caused you not to succeed. Why didn’t you meet your goal?**

YOUR ACTION PLAN

1

2

3

SOLUTIONS

What ***specific*** things do you need to do in the next unit to achieve your goal for the next test (i.e. doing HW, review notes)

NOTE: **If you earned 100%, give advice to your classmates on how you did that and what things they should do to get that**

CAUSES

What did you do to make these your strongest? (Think about your behavior, your work completion, if you asked for help, “It was easy” isn’t acceptable!)

Why were these your lowest areas? (Think about your behavior in class that day, your HW completion, if you asked questions. “It was hard” isn’t acceptable!)

GOAL: What is your SMART goal for your next test! (Specific, Measurable, Attainable, Realistic and Timely)

RESULTS

What areas were the easiest for you?

What areas can you grow in?

SUPPORT

How can **Ms. Francois** help you have a successful Action Plan and hold you accountable?